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Chapter 17 Thermochemistry Worksheet Answers

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Chapter 17 Thermochemistry187 10. Complete the enthalpy diagram for the combustion of natural gas. Use the thermochemical equation in the first paragraph on page 517 as a guide. SECTION 17.3 HEAT IN CHANGES OF STATE (pages 520-526) This section explains heat transfers that occur during melting, freezing, boiling, and condensing.

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)

THERMOCHEMISTRY SECTION 17.1 THE FLOW OF ENERGY-HEAT AND WORK(pages505-510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy Transformations(page505)

THERMOCHEMISTRY

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As liquids absorb heat at their boiling points, the temperature remains constant while they vaporize. true ____ Chapter 17 Thermochemistry 187 05_Chem_GRSW_Ch17.SE/TE 6/11/04 3:49 PM Page 188 Name ____ Date ____ Class ____ CHAPTER 17, Thermochemistry (continued) Use the heating curve for water shown below to answer Questions 5, 6, and 7.

section 17.1 the flow of energy heat and work

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Chapter 17 Thermochemistry429. Section Review. Objectives. •Explain the relationship between energy, heat, and work. •Distinguish between exothermic and endothermic processes. •Distinguish between heat capacity and specific heat. Vocabulary Key Equations and Relationships. •1 Calorie 1 kilocalorie 1000 calories. •1 J 0.2390 cal and 4.184 J 1 cal.

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