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Chemical Reactions A chemical reaction is a change in which one or more substances are converted into new substances Review: a substance is an element or a compound Chemical Reactions The reactants are the substances that exist before the reaction The products are the substances that exist after the reaction Conservation of Mass Conservation of ...

Chapter 21 Chemical Reactions - PC\|MAC

chemical reaction. is a change in which one or more substances are converted into.... reactants. the substances that react. 21 terms. tomsteger33. Chemical Reactions- Chapter 21. chemical reaction. reactants.

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Chemical reaction A process during which chemical bonds between atoms are broken and new ones are formed. Producing one or more different substances.

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Introduction to Chapter 21 There are many different types of reactions that occur around you all of the time. For example, when you breathe you take in oxygen, which reacts with the sugar in your cells. This produces carbon dioxide and water.

Changes in Matter Chapter 21 Types of

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A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

Answer Key Chapter 21 - Chemistry 2e | OpenStax

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Glencoe Physical Science Chapter 21: Chemical Reactions ...

chapter 21 chemical reaction. STUDY. PLAY. oxidation. the lose of electrons. reduction. the gain of electrons. combustion reaction. occurs when a substance reacts with oxygen to produce energy in the form of heat and light. catalyst. a substance that speeds up a chemical reaction without being permanently changed itself.

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Classifying Chemical Reactions. classifying_chemical_reactions_2.docx: File Size: 691 kb: File Type: docx: Download File. Powered by Create your own unique website with customizable templates.

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___ is a substance that speeds up a chemical reaction without being permanently changed itself.
answer choices . catalyst. inhibitor. exotherm ...

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Chapter 21: Chemical Reactions Unit 5: Diversity of Matter Table of Contents 21.3: Classifying Chemical Reactions 21.1: Chemical Changes 21.2: Chemical Equations 21 21.4: Chemical Reactions and Energy •Chemical reactions are taking place all around you and even within you. •A chemical reaction is a

Table of Contents 21

Chapter 21 Chemical Reactions; LHS M. • 40 cards. What is a chemical reaction? It is a change in which one or more substances are converted into new substances. What are reactants? Reactants are the substances that react during a chemical reaction. They are listed on the LEFT side of the arrow. ...

Chapter 21 Chemical Reactions - Physical Science with ...

3. reaction force 4. action force 5. The force also will be 500 N because action-reaction forces are equal and opposite. 6. $p = m v = 2 \text{ kg } 10 \text{ m/s} = 20 \text{ kg} \cdot \text{m/s}$ 7. $p = m v = 2000 \text{ kg } 10 \text{ m/s} = 20,000 \text{ kg} \cdot \text{m/s}$ 8. the 2000-kg truck because it has a greater mass Chapter 4 1. energy 2. potential 3. kinetic 4. gravitational 5. speed Section 1 ...

Study Guide and Reinforcement - Answer Key

The CaSiO_3 slag is less dense than the molten iron, so it can easily be separated. Also, the floating slag layer creates a barrier that prevents the molten iron from exposure to O_2 , which would oxidize the Fe back to Fe_2O_3 .

Answer Key Chapter 19 - Chemistry 2e | OpenStax

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