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Thermodynamics:
Chapter-5 Problems. A
rigid tank of volume 10 m^3 contains
superheated steam at
 1 MPa and $400 \text{ }^\circ\text{C}$.
Due to heat loss to the
outside atmosphere,
the tank gradually
cools down to the
atmospheric
temperature of $25 \text{ }^\circ\text{C}$.

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Determine (a) the heat transfer and (b) the entropy generated in the system's universe during this cooling process.

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Solution Water is found
to move through a pipe
at a speed of 9m/s
where the temperature
of the water at the
entrance is found to be
 C , and pressure 180
 kPa where the pipe
diameter is 0.50m .

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Chapter 5: Homework Solution - Thermodynamics

Chapter 5 The Second
Law of

Thermodynamics In
this chapter we
consider a more
abstract approach to
heat engine,
refrigerator and heat
pump cycles, in an
attempt to determine if
they are feasible, and
to obtain the limiting
maximum performance

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available for these
cycles.

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in this set (50) A
reversible heat pump
cycle operates
between cold and hot

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thermal reservoirs at 300 degrees C and 500 degrees C, respectively. The coefficient of performance is closely (a) 1.5, (b) 3.87, (c) 2.87, (d ...

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N.F.E.E Applications 5.
A closed rigid system has a volume of 85 litres contains steam at 2 bar and dryness fraction of 0.9. Calculate the quantity of heat which must be removed from the system in order to reduce the pressure to 1.0 bar.

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The Kelvin-Planck
limitation applies only
to heat engines;
engines that receive
heat and convert some
of it to work. Chapter 5
The Second Law of
Thermodynamics. 5-2.

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5-17 The power output and thermal efficiency of a power plant are given.

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Chapter 5 THE SECOND LAW OF THERMODYNAMICS

1-1 Chapter 1

INTRODUCTION AND
BASIC CONCEPTS

Thermodynamics 1-1C

Classical

thermodynamics is based on experimental observations whereas statistical

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thermodynamics is based on the average behavior of large groups of particles.

1-2C On a downhill road the potential energy of the bicyclist is being converted to kinetic energy, and thus the bicyclist picks up speed.

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Chapter 3: Homework

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Solution A rigid container has volume of , and holds steam at C. 1/4 of the volume is in liquid point and the remaining at vapor form. Determine the pressure of the steam, and quality of the saturated mixture, and density of the mixture.

Chapter 3: Homework Solution - Thermodynamics

Chapter 5:

Thermodynamic

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Processes and Cycles
5-6) This problem examines the Rankine heat engine introduced in Figure 5-5. Saturated steam at $T = 250^{\circ}\text{C}$ enters the turbine and the condenser operates at $T = 40^{\circ}\text{C}$.

Chapter 5:

Thermodynamic

Processes and

Cycles

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step solutions for your
textbooks written by
Bartleby experts! An
inflated balloon is put
in a refrigerator, and it
shrinks.

**An inflated balloon
is put in a
refrigerator, and it**

...

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Thermodynamics given here will help students to go through all the exercises and improve their ability to tackle all the questions related to the chapter. This is an important unit from an examination point of view. The chapter also focuses on enabling students to learn about fundamental physical properties like entropy, temperature, and energy.

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