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stoichiometry. The study of quantitative relationships between the amounts of reactants used and products formed by a chemical reaction; is based on the law of conservation of mass. mole ratio. In a balanced equation, the ratio between the numbers of moles of any two substances.

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209 Stoichiometry Stoichiometry CHAPTER 11 SOLUTIONS
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Practice Problems pages 371–372 1. Interpret the following
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mass. Show that the law of conservation of mass is

Stoichiometry Stoichiometry

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Activity. Introducing Stoichiometry
Background
A. chemical equation describes the change that occurs when the reactants are converted into products. The balanced equation provides the ratio between the amounts of each one of the chemicals involved.

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