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114 Guided Reading and Study Workbook CHAPTER 11. Chemical Reactions(continued) 9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction. Hydrochloric acid reacts with zinc to produce aqueous zinc(II) chloride and hydrogen gas. Balancing Chemical Equations (pages 324–328) 10.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

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Section 2–4 Chemical Reactions and Enzymes (pages 49–53) This section describes what happens to chemical bonds during chemical reactions. It also explains how energy changes affect whether a chemical reaction will occur and describes the importance of enzymes to living things. Chemical Reactions (page 49) 1. What is a chemical reaction?

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92 Guided Reading and Study Workbook CHAPTER 10.Chemical Quantities(continued) 9. Complete the table about representative particles and moles. The Mass of a Mole of an Element (pages 293–294) 10. What is the atomic mass of an element? 11. Circle the letter of the phrase that completes this sentence correctly. The atomic masses of all elements

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287–296)

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