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Ans: In solid state, the ions cannot move, they are held by strong electrostatic forces of attraction. So, ionic solids do not conduct electricity in solid state. However, in the molten state, they dissociate to give free ions and hence conduct electricity. 1.9. What type of solids are electrical conductors, malleable and ductile?

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volume, rigidity, high density, low compressibility The constituent particles (atoms, molecules or ions) are closely

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