

Thermodynamics Multiple Choice Questions Answers

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Thermodynamics Multiple Choice Questions Answers

This set of Thermodynamics Multiple Choice Questions & Answers focuses on "Maximum Work in a Reversible Process". 1. For a process from state 1 to state 2, heat transfer in a reversible process is given by a) $Q_{\text{for reversible}} = (T_0) * (S_1 - S_2)$

Thermodynamics Multiple Choice Questions and Answers

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Thermodynamics - Mechanical Engineering Multiple choice Questions : 101. Addition of heat at constant pressure to a gas results in (a) raising its temperature (b) raising its pressure (c) raising its volume (d) raising its temperature and doing external work (e) doing external work. Ans: d. 102. Carnot cycle has maximum efficiency for (a) reversible engine

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Thermodynamics Multiple Choice Questions and Answers. 1. The latent heat of steam at atmospheric pressure is..... 2. The efficiency of Ericsson cycle is.....Carnot cycle. D.

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(A) First Law of Thermodynamics (B) Second Law of Thermodynamics (C) Mechanical equivalent of heat (D) Zeroth Law of Thermodynamics (E) Thermal expansion of solids 2. An aluminum plate has a circular hole. If the temperature of the plate increases, what happens to the size of the hole?

Thermodynamics - NJCTL

Multiple choice questions with one or more answers Question 1 It is known that curves A, B, C are Isobaric, Isothermal, Adiabatic process then when one is correct (a) A - Adiabatic, B - Isothermal,, C - Isobaric (b) A- Isothermal,, B - Adiabatic, C - Isobaric (c) A - Isobaric, B - Isothermal C - Adiabatic (d) None of these Solution

Multiple Choice questions on heat and thermodynamics for ...

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Thermodynamics ⇒ Mechanical Engineering Multiple Choice ...

ANSWER: always greater than zero . Explanation: It is impossible for a system to undergo a reversible isothermal process without transfer of heat. To make this process possible, the temperature required is an absolute zero temperature and the second law of thermodynamics is also violated.

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